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SIX NEW DITERPENE ISONITRILES FROM THE SPONGE ACANTHELLA CAVERNOSA

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ABSTRACT.—Six new diterpene isonitriles, $10-\varphi i$ -isokalihinol F [1], $10-\varphi i$ -isokalihinol H [2], $1-\varphi i$ -kalihinene [4], 15-isothiocyanato- $1-\varphi i$ -kalihinene [5], $1,10-di\varphi i$ -kalihinene [6], and kalihipyran [7] were isolated, together with the known compound isokalihinol B [3], from *Acanthella cavernosa* collected in the Seychelles. The structures of the new diterpenes were elucidated by interpretation of spectral data. The ring system of kalihipyran [7] differs from that of the kalihinols and kalihinenes.

The kalihinols and related compounds are diterpenoids from sponges of the genus *Acanthella* that contain two, and often three, isonitrile functionalities (1-6). Members of this series of compounds have been reported to have antimicrobial (1-3), antifungal (1-3), cytotoxic (5), and anthelmintic (4,6) properties. We wish to report the isolation of seven diterpene isonitriles [1-7] from a specimen of *Acanthella cavernosa* Dendy, 1922 (Axinellida, Dictyonellidae), collected in the Seychelles.¹ One of these diterpene isonitriles [**3**], was previously isolated from *A. klethra* (5).

RESULTS AND DISCUSSION

The EtOAc-soluble material from a MeOH extract of a small sample of the sponge A. cavernosa was fractionated by chromatography on Si gel to obtain fractions rich in terpenoids. Further separation of these fractions by hplc led to the isolation of 10-epi-isokalihinol F[1](0.055% dry wt), 10-epi-isokalihinol H[2](0.02% dry wt), isokalihinol B[3](0.006% dry wt), 1-epi-kalihinene [4](0.087% dry wt), 15-isothiocyanato-1-epi-kalihinene [5](0.016% dry wt), 1,10-diepi-kalihinene [6](0.031% dry wt), and kalihipyran [7](0.019% dry wt).

10-epi-Isokalihinol F $\{1\}$ was isolated as a white microcrystalline solid, mp 117°. High-resolution mass spectrometry established a molecular formula of $C_{23}H_{33}N_3O_2$, isomeric with the known compounds kalihinol F [8] and isokalihinol F [9]. The ir spectrum contained bands at 3280 and 2135 cm⁻¹ that were indicative of hydroxyl and isonitrile groups. Comparison of the 13 C-nmr spectrum of **1**, which was assigned with the aid of the HMBC experiment, with those of the known kalihinols revealed a very close similarity with that of isokalihinol F [9] but not with that of kalihinol F [8]. In particular, the signal at δ 76.7 (d) in the ¹³C-nmr spectrum required the presence of a secondary alcohol at C-5 while the chemical shifts of signals assigned to the C-11 to C-18 portion of the molecule were almost identical to those of other kalihinols. Clearly, the only difference between 10-epi-isokalihinol F [1] and isokalihinol F [9] involved the stereochemistry about the bicyclic ring system. The 1 H-nmr spectrum² was assigned by interpretation of the COSY experiment that revealed two contiguous spin systems, one of which corresponded to the H-12 to H-14 moiety. The coupling constants for the H-6 signal at δ 1.71 (q, 1H, J=10 Hz) required that H-1, H-5, H-6, and H-7 all be axial in a trans-decalin ring system. The stereochemistry at C-4 and C-10 was defined by the 13 C-nmr chemical shifts of the C-19 and C-20 methyl signals at δ 19.7 (q) and 27.7 (q)

¹A preliminary report (7) that this sponge [90–075] contained bromophakellin was incorrect.

 $^{^{2}}$ Many of the nmr spectra were recorded in C₆D₆ solution because the signals were better resolved in that solvent.



(cf., δ 19.9 and 20.3 in [9]), which requires axial and equatorial methyl groups, respectively. The stereochemistry about the tetrahydrofuran ring was defined by the observation of a strong nOe of the H-14 signal on irradiation of the CH₃-18 signal. It has normally been assumed that the relative stereochemistry at the C-7, C-11 junction is the same as determined by X-ray analysis for other members of this series; support for this assumption was provided in the case of compound 1 by the observation of enhancements of the H-6 and 5-OH signals on irradiation of the CH₃-18 signal, which were predicted by examination of a molecular model generated by the PC Model program.³

10-epi-Isokalihinol H [2] was isolated as a colorless oil of molecular formula $C_{23}H_{33}N_3O_2S$. The molecular formula suggested the presence of an isothiocyanate in place of one of the isonitrile groups in [1] and the ir spectrum contained bands at 3280, 2135, and 2085 cm⁻¹ that were assigned to hydroxyl, isonitrile, and isothiocyanate groups, respectively. Comparison of the ¹H and ¹³C-nmr spectra of compounds 1 and 2 revealed considerable similarity and suggested that one of the isonitrile groups in 1 was replaced by an isothiocyanate group in 2. The isothiocyanate group was placed at C-10 because, when compared with the ¹H-nmr spectrum of 1, the H-1 and CH₃-20 signals in the ¹H-nmr spectrum did not show broadening due to coupling to the nitrogen in an isonitrile group. The stereochemistry at C-10 was determined by observation of nOes of the H-1 (8%) and H-2_{eq} (7.5%) signals on irradiation of the CH₃-20 signal.

Isokalihinol B [3] was isolated as a very minor constituent of molecular formula $C_{23}H_{33}ClN_2O_2$. Although it was difficult to compare our nmr spectral data recorded in C_6D_6 solution (9) with those reported for a CDCl₃ solution (5), interpretation of our

 $^{^{3}}$ The PC Model program (Serena Software) predicts a high barrier to rotation about the C-7, C-11 bond and a conformation of **1** that places CH₃-18 close to H-6 and 5-OH.

spectral data suggested that the compounds were the same. In particular, the chlorine must be at C-15 because the ¹H-nmr signals for both the 16- and 17-methyl groups were sharpened, and the chemical shifts of the ¹³C-nmr signals for both the 19- and 20-methyl groups are in the correct range for axial methyl groups. It is interesting to note that our sample of isokalihinol B [**3**] has a relatively strong optical rotation, $[\alpha]D + 56.7^{\circ}$, while the literature sample was racemic.

1-epi-Kalihinene [4] was isolated as a colorless oil of molecular formula $C_{22}H_{32}N_2O$, isomeric with kalihinene [10], the structure of which was determined by X-ray analysis (5). The ir spectrum contained a band at 2130 cm⁻¹ due to the isonitrile functionality but there was no hydroxyl band present. The ¹H-nmr spectrum (in C_6D_6), which was assigned by interpretation of the COSY experiment, contained an olefinic proton signal at δ 6.22 (br s, 1 H) that was assigned to H-5. The H-5 signal was not coupled to the H-6 signal at δ 1.76 (br t, 1 H, J=11 Hz) because, according to models, the dihedral angle between H-5 and H-6 is about 90°. The H-6 signal must therefore be coupled to two other axial proton signals (H-1 and H-7) in a *trans*-decalin ring system. The methyl group at C-10 must be axial because irradiation of the H-8_{ax} signal at δ 0.58 caused enhancement of both the H-6 and Me-20 signals. The stereochemistry about the tetrahydrofuran ring was assigned by observing an enhancement of the H-14 signal at δ 3.43 on irradiation of the Me-18 signal at δ 0.73. It must be assumed that the relative stereochemistry at C-7 and C-11 is the same as in all other members of this series.

15-Isothiocyanato-1-epi-kalihinene [5] was isolated as a colorless oil of molecular formula $C_{22}H_{32}N_2OS$. The ir spectrum contains bands at 2125 (-N=C) and 2095 (-NCS) cm⁻¹, indicating that the sulfur atom was incorporated in an isothiocyanate group. The ¹H- and ¹³C-nmr spectra were almost identical to those of 4 except in the region of C-15, where the isothiocyanate group replaces an isonitrile. The ¹H-nmr signals assigned to Me-16 and Me-17 at δ 1.04 and 0.79 are sharp, indicating isothiocyanate substitution at C-15, while the Me-20 signal at δ 0.90 is broad due to coupling with the C-10 isonitrile nitrogen.

1,10-diepi-Kalihinene [6] was isolated as a colorless oil of molecular formula $C_{22}H_{32}N_2O$, isomeric with kalihinene and 1-epi-kalihinene [4]. The ir spectrum again contained an isonitrile band at 2130 cm⁻¹. The ¹H-nmr spectrum of 6 was similar in many respects to that of 4 except that the H-6 signal at δ 2.34 (br t, 1 H, J=12 Hz) was shifted 0.67 ppm downfield from its position in 4 while keeping the same coupling constants. This indicated that the substitution pattern at C-1, C-6, and C-7 in the *trans*-decalin ring system remained the same but that epimerization had occurred at C-10. This was supported by the downfield shift of the Me-20 signal in the ¹³C-nmr spectrum from δ 20.0 in 4 to 27.5 in 6 and an nOe experiment in which irradiation of the Me-20 signal caused enhancements of the H-1 and H-2_{e0} signals.

Kalihipyran [7] was isolated as a colorless oil. The molecular formula, $C_{21}H_{29}NO$, suggested a diterpene with only one isonitrile group and two more unsaturation equivalents than the kalihinenes. The presence of the isonitrile group was confirmed by the observation of an ir band at 2130 cm⁻¹. The ¹³C-nmr spectrum revealed the presence of six olefinic signals, of which two were trisubstituted and the third was a disubstituted methylene. Kalihipyran [7] is therefore tricyclic. Examination of the ¹H- and ¹³C-nmr spectra strongly suggested that the *trans*-decalin ring system found in 1-*epi*-kalihinene [4] was present but that there was a very different monocyclic eight-carbon unit attached at C-7. The COSY experiment revealed strong coupling between the olefinic signal at δ 5.34 (br dd, 1 H, J=5.5 and 3.5 Hz, H-12) and methylene proton signals at δ 2.17 (br ddd, 1 H, J=17, 10.5, and 3.5 Hz, H-13) and 1.88 (br ddd, J=17, 5.5, and 3.5 Hz, H-13), that were in turn coupled to an oxymethine signal at δ 3.84 (br dd, 1 H, J=10.5

and 3.5 Hz, H-14). Weaker allylic couplings were observed between the olefinic proton signal at δ 5.34 and two oxymethine proton signals at δ 4.06 (br d, 1H, J=16 Hz, H-18) and 3.99 (br d, 1H, J=16 Hz, H-18), between the oxymethine proton signal at δ 3.84 and the olefinic methylene signals at δ 5.17 (br s, 1H, H-16) and 4.89 (br s, 1 H, H-16) and between the methyl signal at δ 1.74 (s, 3H) and the same olefinic methylene signals. Weak homoallylic coupling between the H-13 and H-18 signals was also noted. Two nOeds experiments provided support for this structure: irradiation of the H-14 signal caused enhancements of the H-16 (δ 4.98), Me-17, and H-12 (δ 4.06) signals and irradiation of the H-12 signal at δ 4.06 caused enhancement of the H-14 and H-5 signals. It was not possible to determine the relative stereochemistry at C-14.

The oxidation pattern observed in kalihipyran [7] is different from that observed in other kalihinols and kalihinenes. The crude extract showed very modest inhibition of *Staphylococcus aureus, Bacillus subtilis,* and *Candida albicans* but, although the activity was sufficient to follow during fractionation, the pure compounds showed only minimal zones of inhibition at 100 μ g/disk.

EXPERIMENTAL

ANIMAL MATERIAL, EXTRACTION AND CHROMATOGRAPHY.—The orange sponge Acanthella cavernosa Dendy, 1922 (SIO invertebrate Collection no. P1136) was collected by hand (-10 m) at Beau Vallon Beach, Mahé, Seychelles, and kept frozen until extraction. The sponge (7.5 g dry weight) was extracted twice with MeOH, the solvent was evaporated, and the residue was partitioned between H₂O and EtOAc. The EtOAc extract (86 mg) was subjected to flash chromatography on Si gel using a solvent gradient from hexane to EtOAc to yield a number of fractions that appeared by ¹H-nmr spectroscopy to contain diterpenes. Fractions eluted with hexane to 10% EtOAc in hexane were combined and separated by hplc on Partisil 10 using 20% EtOAc in hexane as eluent to obtain 1-epi-kalihinene [4] (2.3 mg), 15-isothiocyanatokalihinene [5] (1.2 mg), 1,10-diepi-kalihinene [6] (3.8 mg), and kalihipyran [7] (1.4 mg). The fraction eluting with 20% EtOAc in hexane was separated under identical hplc conditions to obtain 1-epi-kalihinene [4] (4.2 mg, 6.7 mg total) and 1,10-diepi-kalihinene [6] (1.1 mg, 4.9 mg total). Fractions eluting with 50–60% EtOAc in hexane were separated by hplc on Si gel using 30% EtOAc in hexane as eluent to obtain 10-epi-isokalihinol F [1] (4.1 mg), 10-epi-isokalihinol H [2] (1.5 mg) and isokalihinol B [3] (0.45 mg).

10-*EPI*-ISOKALIHINOL F [1].—Mp 117° (CH₂Cl₂); $[\alpha]_D - 21.2°$ (z=0.41, CH₂Cl₂); ir (KBr) ν max 3280, 2975, 2940, 2135, 1685, 1450, 1380, 1030 cm⁻¹; ¹H-nmr (CDCl₃) δ 6.41 (1H, d, J=3.5 Hz, OH-5), 3.89 (1H, br m, H-14), 3.63 (1H, br d, J=10 Hz, H-5), 2.17 (1H, m, H-13), 2.10 (1H, dt, J=9 and 3 Hz, H-3_{eq}), 2.04 (1H, m, H-13), 1.96 (1H, br d, J=13 Hz, H-8_{eq}), 1.91 (2H, m, H-12), 1.81 (1H, m, H-3_a), 1.80 (1H, dq, J=10 and 3.5 Hz, H-2_{eq}), 1.71 (1H, q, J=10 Hz, H-6), 1.66 (1H, m, H-9_{eq}), 1.61 (1H, dddd, J=13, 12, 10, and 3 Hz, H-8_a), 1.54 (1H, td, J=10 and 3 Hz, H-7), 1.48 (3H, br s, H-19), 1.47 (3H, br t, J=10 Hz, H-16), 1.44 (1H, m, H-2_a), 1.41 (3H, br s, H-20), 1.36 (3H, br s, H-17), 1.34 (3H, s, H-18), 1.20 (1H, br t, J=10 Hz, H-1); ¹³C nmr, see Table 1; ms, m/z 384 [MH]⁺, 356 [M-HCN]⁺, 327 [M-2HCN]⁺, 302 [M-3HCN]⁺, 285 [M-3HCN-OH]⁺; hrms, m/z 383.2545 (C₂₃H₃₃N₃O₂ requires 383.2572).

10-*EPI*-ISOKALIHINOL H [2].—Colorless oil; $[\alpha]D - 32.1^{\circ} (c=0.17, CH_2Cl_2)$; uv (MeOH) λ max 244 nm (ϵ 1265); ir (neat) ν max 3360, 2940, 2870, 2135, 2085 (br), 1455, 1380, 1180, 885 cm⁻¹; ¹H nmr (C₆D₆) δ 6.22 (1H, d, J=4 Hz, OH-5), 3.51 (1H, dd, J=11 and 3.5 Hz, H-5), 2.94 (1H, br m, H-14), 1.65 (1H, dt, J=13.5 and 3.5 Hz, H-3_{eq}), 1.56 (1H, q, J=11 Hz, H-6), 1.46 (3H, br s, H-19), 1.22 (1H, br t, J=11.5 Hz, H-7), 1.08 (3H, br s, H-16), 1.01 (3H, s, H-18), 0.91 (1H, qd, J=12 and 3 Hz, H-2_{ex}), 0.79 (1H, qd, J=12 and 3 Hz, H-8_{ex}), 0.71 (3H, s, H-20), 0.63 (3H, br s, H-17), 0.37 (1H, td, J=11 and 4.5 Hz, H-1); ¹³C nmr, see Table 1; hrfabms, m/z 548.1348 [M+Cs]⁺ (C₂₃H₃₃N₃O₂SCs requires 548.1348).

ISOKALIHINOL B [3].—Colorless oil; $[\alpha]D + 56.7^{\circ}(c=0.03, CH_2Cl_2)$; ir (near) ν max 3350, 2960, 2145, 1380, 1130, 895 cm⁻¹; ¹H nmr (C₆D₆) δ 6.31 (1H, d, J=4 Hz, OH-5), 3.41 (1H, dd, J=9.5 and 4.5 Hz, H-14), 3.40 (1H, dd, J=9.5 and 4 Hz, H-5), 1.65 (1H, m, H-13), 1.64 (1H, dt, J=13.5 and 3.5 Hz, H-9_{eq}), 1.58 (1H, dq, J=13.5 and 3.5 Hz, H-8_{eq}), 1.50 (1H, dt, J=13 and 3.5 Hz, H-3_{eq}), 1.46 (1H, m, H-13), 1.40 (1H, br t, J=13 Hz, H-9_{eq}), 1.39 (1H, br t, J=14 Hz, H-3_{eq}), 1.37 (1H, m, H-12), 1.35 (3H, s, H-16), 1.25 (1H, m, H-12), 1.23 (3H, br t, J=1.5 Hz, H-19), 1.15 (3H, s, H-18), 1.07 (1H, ddd, J=12.5, 9.5, and 3.5 Hz, H-1), 0.94 (1H, m, H-2_{eq}), 0.92 (1H, dt, J=1.5 Hz, H-20), 0.64 (1H, tdd, J=14, 12, and

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| Carbon | Compound | | | | | | |
|--------|---|--|--|---|---|---|--|
| | 1' | 2' | 3' | 4 ^b | 5' | 6 ⁵ | 7 |
| 1 | 47.0 (d) 21.0 (t) 36.8 (t) 61.6° 76.7 (d) 43.4 (d) 53.9 (d) 26.5 (t) 39.0 (t) 60.8° 88.2 (s) 39.2 (t) 25.0 (t) 82.4 (d) 60.1° 25.7 (q) 18.4 (q) 19.6 (q) 27.7 (q) 151.9° | $\begin{array}{c} 48.7 \text{ (d)} \\ 21.3 \text{ (t)} \\ 36.9 \text{ (t)} \\ -h \\ 76.8 \text{ (d)} \\ 43.6 \text{ (d)} \\ 53.8 \text{ (d)} \\ 26.9 \text{ (t)} \\ 39.8^{d} \text{ (t)} \\ 64.2 \text{ (s)} \\ 87.9 \text{ (s)} \\ 38.9^{d} \text{ (t)} \\ 64.2 \text{ (s)} \\ 87.9 \text{ (s)} \\ 38.9^{d} \text{ (t)} \\ 25.0 \text{ (t)} \\ 82.3 \text{ (d)} \\ -25.5^{c} \text{ (q)} \\ 25.8^{c} \text{ (q)} \\ 18.3 \text{ (q)} \\ 19.7 \text{ (q)} \\ 27.3 \text{ (g)} \\ -h \\ -h \end{array}$ | $\begin{array}{c} 46.9 \text{ (d)} \\ 21.1 \text{ (t)} \\ 36.7 \text{ (t)} \\ 61.5 \text{ (f)} \\ 76.7 \text{ (d)} \\ 42.7 \text{ (d)} \\ 53.6 \text{ (d)} \\ 25.4^d \text{ (t)} \\ 40.3^e \text{ (t)} \\ 59.5^e \\ 87.2 \text{ (s)} \\ 38.9^e \text{ (t)} \\ 25.4^d \text{ (t)} \\ 83.9 \text{ (d)} \\ 71.0 \text{ (s)} \\ 28.9^f \text{ (q)} \\ 27.7^f \text{ (q)} \\ 17.8 \text{ (q)} \\ 19.9^g \text{ (q)} \\ 19.9^g \text{ (q)} \\ - b \end{array}$ | $\begin{array}{c} 47.3 (d) \\ 23.6 (t) \\ 30.3 (t) \\ 133.2 (s) \\ 124.2 (d) \\ 38.3 (d) \\ 51.3 (d) \\ 25.1^{d} (t) \\ 37.9 (t) \\ 60.4^{c} \\ 87.1 (s) \\ 40.5 (t) \\ 25.5^{d} (t) \\ 82.6 (d) \\ 60.3^{c} \\ 25.0^{c} (q) \\ 18.4 (q) \\ 23.8 (q) \\ 20.0 (q) \\ 152.1^{c} \\ 153.9^{c} \end{array}$ | $\begin{array}{c} 47.5 \text{ (d)} \\ 24.1 \text{ (t)} \\ 30.4 \text{ (t)} \\ 133.1 \text{ (s)} \\ 124.6 \text{ (d)} \\ 38.6 \text{ (d)} \\ 51.4 \text{ (d)} \\ 25.2^{d} \text{ (t)} \\ 37.8 \text{ (t)} \\ 60.2^{c} \\ 87.3 \text{ (s)} \\ 40.7 \text{ (t)} \\ 25.6^{d} \text{ (t)} \\ 83.5 \text{ (d)} \\ 63.7 \text{ (s)} \\ 25.4^{e} \text{ (q)} \\ 25.9^{e} \text{ (q)} \\ 19.9 \text{ (q)} \\ 23.9 \text{ (q)} \\ 18.3 \text{ (q)} \\ 154.8^{c} \\ \end{array}$ | $\begin{array}{c} 46.6 \text{ (d)} \\ 23.2 \text{ (t)} \\ 30.1 \text{ (t)} \\ 132.3 \text{ (s)} \\ 124.8 \text{ (d)} \\ 38.5 \text{ (d)} \\ 51.3 \text{ (d)} \\ 25.2^{d} \text{ (t)} \\ 38.0 \text{ (t)} \\ 60.5^{c} \\ 87.5 \text{ (s)} \\ 39.7 \text{ (t)} \\ 25.4^{d} \text{ (t)} \\ 82.7 \text{ (d)} \\ 60.5^{c} \\ 25.4^{d} \text{ (q)} \\ 26.2^{d} \text{ (q)} \\ 18.4 \text{ (q)} \\ 23.7 \text{ (q)} \\ 27.5 \text{ (q)} \\ - 5 \\ - 5 \end{array}$ | $\begin{array}{c} 47.2 (d) \\ 23.4 (t) \\ 30.8 (t) \\ 135.4 (s) \\ 121.9 (d) \\ 38.3 (d) \\ 47.9 (d) \\ 28.9^d (t) \\ 40.5 (t) \\ 60.3^c \\ 145.2 (s) \\ 120.0 (d) \\ 29.6^d (t) \\ 75.8 (d) \\ 138.4 (s) \\ 110.6 (t) \\ 19.9 (q) \\ 66.9 (t) \\ 23.3 (q) \\ 18.8 (q) \\ 152.4^c \\ \underline{}^b \end{array}$ |
| NC | 155.5° | h | | <u> </u> | h | h | |

TABLE 1. ¹³C-Nmr Data [50 MHz, δ (mult.)] for Diterpenes 1–7.

 $^{\mathbf{C}_{6}}\mathbf{D}_{6}$ solution.

^bCDCl₃ solution.

Signal appears as a broad 1:1:1 triplet due to coupling with the isonitrile nitrogen.

^dSignals may be interchanged.

^hSignal not observed.

3.5 Hz, H-8,), 0.47 (1H, tdd, J=14, 12.5, and 3.5 Hz, H-2,); ¹³C nmr, see Table 1; ms, m/z 392/394 [M]⁺, 366/368 [M-CN]⁺, 357 [M-Cl]⁺, 339/341 [M-CN-HCN]⁺, 330 [M-Cl-HCN]⁺, 303 [M-Cl-HCN]⁺; hrfabms, m/z 525.1264 [M+Cs]⁺ (C₂₂H₃₃N₂O³⁵Cl requires 525.1285).

1-*EPI*-KALIHINENE [4].—Colorless oil; $[\alpha]D - 171.5^{\circ}(c=0.431, CH_2Cl_2)$; ir (neat) ν max 2965, 2875, 2130, 1670, 1450, 1380, 1085, 1040, 820 cm⁻¹; ¹H nmr (C₆D₆) δ 6.22 (1H, br s, H-5), 3.43 (1H, br m, H-14), 2.05 (1H, ddt, J=12, 6, and 2 Hz, H-2_{eq}), 1.85 (1H, br t, J=17 Hz, H-3_{ar}), 1.74 (1H, dd, J=17 and 5 Hz, H-3_{eq}), 1.67 (1H, br t, J=11 Hz, H-6), 1.65 (3H, s, H-19), 1.62 (1H, dt, J=13 and 3.5 Hz, H-9_{eq}), 1.55 (1H, m, H-9_{ar}), 1.53 (2H, m, H-13), 1.39 (1H, q, J=11 Hz, H-12), 1.35 (1H, br t, J=11 Hz, H-2_{eq}), 1.21 (1H, ddd, J=12.5, 11, and 3.5 Hz, H-7_{ar}), 1.17 (1H, ddd, J=11.5, 8, and 4 Hz, H-12), 1.12 (3H, br s, H-16), 1.06 (1H, dq, J=14 and 3.5 Hz, H-8_{eq}), 1.05 (1H, m, H-2_{ar}), 0.93 (3H, br s, H-17), 0.91 (3H, br t, J=1.5 Hz, H-20), 0.73 (3H, s, H-18), 0.58 (1H, dddd, J=14, 13, 12, and 4 Hz, H-8_{ar}); ¹³C nmr, see Table 1; ms m/z 340 [M]⁺, 325 [M-CH₃]⁺, 313 [M-HCN]⁺, 298 [M-CH₃-HCN]⁺ 286 [M-2HCN]⁺; hrms, m/z 340.2500 (C₂₂H₃₂N₂O requires 340.2515).

15-ISOTHIOCYANATO-1-*EPI*-KALIHINENE [**5**].—Colorless oil; $[\alpha]D - 149^{\circ}$ (c=0.132, CH₂Cl₂); uv (MeOH) λ max 245 nm (ϵ 1206); ir (neat) ν max (neat) 2965, 2935, 2875, 2125, 2095 (br), 1450, 1380, 1085, 1037 cm⁻¹; ¹H nmr (C₆D₆) δ 6.17 (1H, br s, H-5), 3.38 (1H, dd, J=8 and 4 Hz, H-14), 2.03 (1H, ddt, J=12, 6, and 2 Hz, H-2_{eq}), 1.81 (1H, br t, J=17 Hz, H-3_x), 1.70 (1H, dd, J=17 and 5 Hz, H-3_{eq}), 1.67–1.63 (2H, m), 1.62 (3H, br s, H-19), 1.45–1.32 (4H, m), 1.26 (1H, ddd, J=14, 10, and 3.5 Hz, H-9_x), 1.16–1.00 (3H, m), 1.04 (3H, s, H-16), 0.90 (3H, br t, J=1.5 Hz, H-20), 0.79 (3H, s, H-17), 0.71 (3H, s, H-18); ¹³C nmr, see Table 1; hrfabms m/z 505.1315 [M+Cs]⁺ (C₂₂H₃₂N₂OSCs requires 505.1290).

1,10-*DIEPI*-KALIHINENE [**6**].—Colorless oil, $[\alpha]D - 66.2^{\circ}$ (c=0.47, CH₂Cl₂); ir (neat) ν max 2925, 2835, 2130, 1455, 1380, 1085, 1040, 890 cm⁻¹; ¹H nmr (C₆D₆) δ 6.38 (1H, br s, H-5), 3.35 (1H, br m, H-14), 2.34 (1H, br t, J=12 Hz, H-6), 1.84–1.74 (2H, m), 1.71 (3H, br s, H-19), 1.65 (1H, dq, J=12 and 3 Hz, H-2_w), 1.58–1.44 (6H, m), 1.41 (1H, qd, J=12 and 3.5 Hz, H-8_w), 1.30 (1H, td, J=11 and 3.5 Hz, H-7_w), 1.24 (1H, dq, J=12 and 3.5 Hz, H-8_w), 1.16 (3H, br s, H-16), 0.97 (3H, br s, H-20), 0.91 (3H, s, H-18), 0.87 (3H, br s, H-17), 0.72 (1H, br td, J=12 and 3 Hz, H-1; ¹³C nmr, see Table 1; ms, m/z 340

 $[M]^+$, 314 $[M-CN]^+$, 288 $[M-2CN]^+$, 287 $[M-CN-HCN]^+$; hrms m/z 340.2528 $(C_{22}H_{32}N_2O$ requires 340.2515).

KALIHIPYRAN [7].—Colorless oil; $[\alpha]D + 104.1^{\circ}$ (c=0.29, CH₂Cl₂); ir (neat) ν max 2935, 2870, 2130, 1655, 1450, 1380, 1090, 900 cm⁻¹; ¹H nmr (CDCl₃) δ 5.34 (1H, br dd, J=5.5 and 3.5 Hz, H-12), 5.30 (1H, s, H-5), 5.17 (1H, br s, H-16), 4.89 (1H, br s, H-16), 4.06 (1H, br d, J=16 Hz, H-18), 3.99 (1H, br d, J=16 Hz, H-18), 3.84 (1H, br dd, J=10.5 and 3.5 Hz, H-14), 2.17 (1H, br ddd, J=17, 10.5, and 3.5 Hz, H-13), 1.98 (1H, ddt, J=13, 6.5, and 2 Hz, H-2), 1.88 (1H, br ddd, J=17, 5.5, and 3.5 Hz, H-13), 1.75 (2H, m, H-3), 1.74 (3H, s, H-17), 1.63 (1H, m, H-8), 1.58 (1H, m, H-9), 1.52 (3H, s, H-19), 1.50 (1H, m, H-6), 1.27 (1H, m, H-7), 1.26 (1H, m, H-1), 1.22 (1H, m, H-9), 1.01 (1H, m, H-2), 0.99 (1H, m, H-8), 0.87 (3H, br t, J=1.5 Hz, H-20); ¹³C nmr, see Table 1; ms, m/z 311 [M]⁺, 284 [M-HCN]⁺; hrfabms m/z 334.2157 [M+Na]⁺ (C₂₁H₂₉NONa requires 334.2147).

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